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<u>Solute</u> – the substance that dissolved (smaller amount)	<u>Solvent</u> - the substance doing the dissolving (bigger amount); water is the universal solvent	A solution is formed when one substance dissolves in another substance. Solute + Solvent = Solution

<u>**Dilute</u>** = too much solvent</u>

<u>Saturated</u> = too much solute that cannot be dissolved any further

Solubility - is the ability of a substance to dissolve (become trapped in) another substance.

Temperature can increase solubility.

Example: hot tea vs. iced tea (with sugar added)